

Solve for convective film coefficient using: Principles of Heat Transfer by Kreith

$$\begin{aligned}
 T_c &:= (273.15 - 35)K & T_c &= 238.15K & kJ &:= 1000J & \text{mbar} &:= 10^{-3}\text{bar} & \mu\text{Pa} &:= 10^{-6}\text{Pa} \\
 Q &:= 240W & L1 &:= 2m & \sigma &:= 0.012\frac{N}{m} & c_{\text{liq}} &:= 2039\frac{J}{\text{kg}\cdot K} \\
 \rho_{\text{liq}} &:= 1096\frac{\text{kg}}{\text{m}^3} & \rho_v &:= 31\frac{\text{kg}}{\text{m}^3} & \mu_{\text{liq}} &:= 178\mu\text{Pa}\cdot\text{s} & \mu_v &:= 12\mu\text{Pa}\cdot\text{s} \\
 k_{\text{liq}} &:= 0.153\frac{W}{\text{m}\cdot K} & k_v &:= 0.013\frac{W}{\text{m}\cdot K} \\
 h_{\text{liq}} &:= 123.05\frac{kJ}{\text{kg}} & h_v &:= 436.23\frac{kJ}{\text{kg}} & \Delta h &:= h_v - h_{\text{liq}} & \Delta h &= 313.18\frac{kJ}{\text{kg}} & \lambda &:= \Delta h \\
 x_i &:= 0.05 & x_o &:= 0.85 & \dot{m} &:= \frac{Q}{(x_o - x_i)\cdot\lambda} & \dot{m} &= 9.579 \times 10^{-4}\frac{\text{kg}}{\text{s}}
 \end{aligned}$$

Tube dimensions  $t := .012\text{in}$  wall thickness

$$\begin{aligned}
 d_o &:= 2.8\text{mm} & d_i &:= d_o - 2\cdot t & A_c &:= \frac{\pi}{4}\cdot d_i^2 & \text{round tube} \\
 D_h &:= d_i & D_h &= 2.19\text{mm} & A_t &:= A_c & P_c &:= \pi\cdot D_h & L1 &= 2m \\
 G1 &:= \frac{\dot{m}}{A_t} & G1 &= 254.209\frac{\text{kg}}{\text{m}^2\cdot\text{s}} \\
 h_{\text{flux}} &:= \frac{240W}{P_c\cdot L1} & h_{\text{flux}} &= 1.744 \times 10^4\frac{W}{\text{m}^2} & T_{\text{sat}} &:= T_c & T_{\text{sat}} &= 238.15K & \text{fluid saturation} & \text{temperature} \\
 k_w &:= 200\frac{W}{\text{m}\cdot K} & \text{aluminum tube wall} & & \text{for } T_{\text{sat}} = 238K & \text{the saturation pressure is } 12.024\text{bar} \\
 P &:= 12.024\text{bar} & P &= 1.202 \times 10^6\text{Pa} & Pr_{\text{liq}} &:= \frac{\mu_{\text{liq}}\cdot c_{\text{liq}}}{k_{\text{liq}}} & Pr_{\text{liq}} &= 2.372 \\
 \alpha_{\text{liq}} &:= \frac{k_{\text{liq}}}{\rho_{\text{liq}}\cdot c_{\text{liq}}} & \alpha_{\text{liq}} &= 6.846 \times 10^{-8}\frac{\text{m}^2}{\text{s}} & \text{thermal diffusivity} & & & & & 
 \end{aligned}$$

Condition for nucleate boiling to occur

$$\Delta T_n := \left( \frac{8\cdot\sigma\cdot h_{\text{flux}}\cdot T_{\text{sat}}}{\lambda\cdot\rho_v\cdot k_{\text{liq}}} \right)^{0.5} \quad \Delta T_n = 0.518K \quad \text{any differential film temp above this}$$

Reference: Kreith

$$i := 0..6$$

$$x := \begin{pmatrix} .05 \\ .1 \\ .3 \\ .5 \\ .6 \\ .7 \\ .85 \end{pmatrix}$$

really method by Chen

this a linear change flow quality along the tube length

$$X_{ttinverse_i} := \left( \frac{x_i}{1-x_i} \right)^{0.9} \cdot \left( \frac{\rho_{liq}}{\rho_v} \right)^{0.5} \cdot \left( \frac{\mu_v}{\mu_{liq}} \right)^{0.1}$$

$$X_{ttinverse} = \begin{pmatrix} 0.321 \\ 0.628 \\ 2.118 \\ 4.54 \\ 6.54 \\ 9.734 \\ 21.632 \end{pmatrix}$$

$$F_{tti} := 2.35 \cdot (X_{ttinverse_i} + 0.213)^{0.736}$$

$$F_{tt} = \begin{pmatrix} 1.481 \\ 2.07 \\ 4.381 \\ 7.402 \\ 9.585 \\ 12.746 \\ 22.742 \end{pmatrix}$$

convective cooling

$$h_{c_i} := 0.023 \cdot \left[ \frac{G1 \cdot (1-x_i) \cdot D_h}{\mu_{liq}} \right]^{0.8} \cdot Pr_{liq}^{0.4} \cdot \frac{k_{liq}}{D_h} \cdot F_{tti}$$

$$h_c = \begin{pmatrix} 2017.35 \\ 2700.713 \\ 4675.668 \\ 6035.495 \\ 6537.599 \\ 6906.327 \\ 7077.704 \end{pmatrix} \cdot \frac{W}{m^2 \cdot K}$$

**$h_c$  is the cooling from forced convection. The next step is to calculate the cooling from evaporation.**

Solving for evaporation requires assuming a temperature difference between the tube wall and the saturation temperature of the fluid. After calculating this contribution we can solve for this temperature difference through the combination of both cooling mechanisms. If the end result agrees with the assumption, the process stops otherwise the process is repeated with a new assumed  $\Delta T_{sat}$

$$Re_{TP_i} := \frac{G1 \cdot (1-x_i) \cdot D_h}{\mu_{liq}} \cdot (F_{tti})^{1.25} \cdot 10^{-4} \quad \text{needed to calculate } S_{tt}$$

$$\text{Re}_{\text{TP}} = \begin{pmatrix} 0.485 \\ 0.699 \\ 1.388 \\ 1.91 \\ 2.11 \\ 2.26 \\ 2.33 \end{pmatrix} \quad \text{Stt}_i := \left[ 1 + 0.12 \cdot (\text{Re}_{\text{TP}_i})^{1.14} \right]^{-0.1} \quad \text{Stt} = \begin{pmatrix} 0.995 \\ 0.992 \\ 0.984 \\ 0.978 \\ 0.976 \\ 0.974 \\ 0.973 \end{pmatrix} \quad \text{essentially constant over the tube length}$$

For the assumed  $\Delta T_{\text{sat}}$  (wall temp-fluid saturation temp) we calculate the value of  $\Delta P_{\text{sat}}$  for CO<sub>2</sub>. This value is the change in saturation pressure corresponding to the change in fluid temperature. At -35C the change in saturation pressure per degree C is 45080Pa, or for 2.5C  $\Delta T_{\text{sat}}$  the change is 112700Pa

$\Delta T_{\text{sat}} := 2.55\text{K}$  temperature difference between wall and fluid, assumed to start the process.

$\Delta P_{\text{sat}} := 114954\text{Pa}$   $\Delta P_{\text{sat}} = 1.15 \times 10^3 \cdot \text{mbar}$  pressure difference for temperature difference of 2C

$$h_{b_0} := .00122 \cdot \left( \frac{k_{\text{liq}}^{0.79} \cdot c_{\text{liq}}^{0.45} \cdot \rho_{\text{liq}}^{0.49}}{\sigma^{0.5} \cdot \mu_{\text{liq}}^{0.29} \cdot \lambda^{0.24} \cdot \rho_v^{0.24}} \right) \cdot \Delta T_{\text{sat}}^{0.24} \cdot \Delta P_{\text{sat}}^{0.75} \cdot \text{Stt}_0 \quad \text{Chen}$$

$$h_{b_0} = 4815.221 \cdot \frac{\text{W}}{\text{m}^2 \cdot \text{K}} \quad h_0 := h_{b_0} + h_{c_0} \quad h_0 = 6.833 \times 10^3 \cdot \frac{\text{W}}{\text{m}^2 \cdot \text{K}}$$

$$\Delta t_{b_0} := \frac{h_{\text{flux}}}{h_0} \quad \Delta t_{b_0} = 2.552\text{K} \quad \Delta T \text{ assumed was } 2.5\text{C}$$

For the next quality 0.1

$\Delta T_{\text{sat}} := 2.4\text{K}$  temperature difference between wall and fluid, assumed to start the process.

$\Delta P_{\text{sat}} := 108192\text{Pa}$   $\Delta P_{\text{sat}} = 1.082 \times 10^3 \cdot \text{mbar}$  pressure difference for temperature difference of 2C

$$h_{b_1} := .00122 \cdot \left( \frac{k_{\text{liq}}^{0.79} \cdot c_{\text{liq}}^{0.45} \cdot \rho_{\text{liq}}^{0.49}}{\sigma^{0.5} \cdot \mu_{\text{liq}}^{0.29} \cdot \lambda^{0.24} \cdot \rho_v^{0.24}} \right) \cdot \Delta T_{\text{sat}}^{0.24} \cdot \Delta P_{\text{sat}}^{0.75} \cdot \text{Stt}_1 \quad \text{Chen}$$

$$h_{b_1} = 4523.196 \cdot \frac{\text{W}}{\text{m}^2 \cdot \text{K}} \quad h_1 := h_{b_1} + h_{c_1} \quad h_1 = 7.224 \times 10^3 \cdot \frac{\text{W}}{\text{m}^2 \cdot \text{K}}$$

$$\Delta t_{b_1} := \frac{h_{\text{flux}}}{h_1} \quad \Delta t_{b_1} = 2.414\text{K} \quad \Delta T \text{ assumed was } 2.4\text{C}$$

For the next quality 0.3

$\Delta T_{sat} := 2.05K$  temperature difference between wall and fluid, assumed to start the process.

$\Delta p_{sat} := 92414Pa$   $\Delta p_{sat} = 924.14 \cdot mbar$  pressure difference for temperature difference of 2C

$$h_{b_2} := .00122 \cdot \left( \frac{k_{liq}^{0.79} \cdot c_{liq}^{0.45} \cdot \rho_{liq}^{0.49}}{\sigma^{0.5} \cdot \mu_{liq}^{0.29} \cdot \lambda^{0.24} \cdot \rho_v^{0.24}} \right) \cdot \Delta T_{sat}^{0.24} \cdot \Delta p_{sat}^{0.75} \cdot Stt_2 \quad \text{Chen}$$

$$h_{b_2} = 3837.293 \cdot \frac{W}{m^2 \cdot K} \quad h_2 := h_{b_2} + h_{c_2} \quad h_2 = 8.513 \times 10^3 \cdot \frac{W}{m^2 \cdot K}$$

$$\Delta t_{b_2} := \frac{h_{flux}}{h_2} \quad \Delta t_{b_2} = 2.048K \quad \Delta T \text{ assumed was } 2.04C$$

For the next quality 0.5

$\Delta T_{sat} := 1.83K$  temperature difference between wall and fluid, assumed to start the process.

$\Delta p_{sat} := 82496Pa$   $\Delta p_{sat} = 824.96 \cdot mbar$  pressure difference for temperature difference of 2C

$$h_{b_3} := .00122 \cdot \left( \frac{k_{liq}^{0.79} \cdot c_{liq}^{0.45} \cdot \rho_{liq}^{0.49}}{\sigma^{0.5} \cdot \mu_{liq}^{0.29} \cdot \lambda^{0.24} \cdot \rho_v^{0.24}} \right) \cdot \Delta T_{sat}^{0.24} \cdot \Delta p_{sat}^{0.75} \cdot Stt_3 \quad \text{Chen}$$

$$h_{b_3} = 3407.788 \cdot \frac{W}{m^2 \cdot K} \quad h_3 := h_{b_3} + h_{c_3} \quad h_3 = 9.443 \times 10^3 \cdot \frac{W}{m^2 \cdot K}$$

$$\Delta t_{b_3} := \frac{h_{flux}}{h_3} \quad \Delta t_{b_3} = 1.847K \quad \Delta T \text{ assumed was } 1.83C$$

For the next quality 0.6

$\Delta T_{sat} := 1.75K$  temperature difference between wall and fluid, assumed to start the process.

$\Delta p_{sat} := 78890Pa$   $\Delta p_{sat} = 788.9\text{-mbar}$  pressure difference for temperature difference of 2C

$$h_{b4} := .00122 \cdot \left( \frac{k_{liq}^{0.79} \cdot c_{liq}^{0.45} \cdot \rho_{liq}^{0.49}}{\sigma^{0.5} \cdot \mu_{liq}^{0.29} \cdot \lambda^{0.24} \cdot \rho_v^{0.24}} \right) \cdot \Delta T_{sat}^{0.24} \cdot \Delta p_{sat}^{0.75} \cdot Stt4$$

$$h_{b4} = 3252.5 \cdot \frac{W}{m^2 \cdot K} \quad h_4 := h_{b4} + h_{c4} \quad h_4 = 9.79 \times 10^3 \cdot \frac{W}{m^2 \cdot K}$$

$$\Delta t_{b4} := \frac{h_{flux}}{h_4} \quad \Delta t_{b4} = 1.781K \quad \Delta T \text{ assumed was } 1.75C$$

For the next quality 0.7

$\Delta T_{sat} := 1.7K$  temperature difference between wall and fluid, assumed to start the process.

$\Delta p_{sat} := 76636Pa$   $\Delta p_{sat} = 766.36\text{-mbar}$  pressure difference for temperature difference of 2C

$$h_{b5} := .00122 \cdot \left( \frac{k_{liq}^{0.79} \cdot c_{liq}^{0.45} \cdot \rho_{liq}^{0.49}}{\sigma^{0.5} \cdot \mu_{liq}^{0.29} \cdot \lambda^{0.24} \cdot \rho_v^{0.24}} \right) \cdot \Delta T_{sat}^{0.24} \cdot \Delta p_{sat}^{0.75} \cdot Stt5$$

$$h_{b5} = 3154.901 \cdot \frac{W}{m^2 \cdot K} \quad h_5 := h_{b5} + h_{c5} \quad h_5 = 1.006 \times 10^4 \cdot \frac{W}{m^2 \cdot K}$$

$$\Delta t_{b5} := \frac{h_{flux}}{h_5} \quad \Delta t_{b5} = 1.733K \quad \Delta T \text{ assumed was } 1.7C$$

For the next quality 0.85

$\Delta T_{sat} := 1.7K$  temperature difference between wall and fluid, assumed to start the process.

$\Delta p_{sat} := 76636Pa$   $\Delta p_{sat} = 766.36\text{-mbar}$  pressure difference for temperature difference of 2C

$$h_{b6} := .00122 \cdot \left( \frac{k_{liq}^{0.79} \cdot c_{liq}^{0.45} \cdot \rho_{liq}^{0.49}}{\sigma^{0.5} \cdot \mu_{liq}^{0.29} \cdot \lambda^{0.24} \cdot \rho_v^{0.24}} \right) \cdot \Delta T_{sat}^{0.24} \cdot \Delta P_{sat}^{0.75} \cdot Stt6$$

$$h_{b6} = 3152.299 \cdot \frac{W}{m^2 \cdot K} \quad h_6 := h_{b6} + h_{c6} \quad h_6 = 1.023 \times 10^4 \cdot \frac{W}{m^2 \cdot K}$$

$$\Delta t_{b6} := \frac{h_{flux}}{h_6} \quad \Delta t_{b6} = 1.705 K \quad \Delta T \text{ assumed was } 1.7C$$